

Organic Chemistry

Common Function Groups

Structures of Some Common Functional Groups

Name	Structure*	Name ending	Example
Alkene (double bond)		-ene	H ₂ C=CH ₂ Ethene
Alkyne (triple bond)		-yne	HC≡CH Ethyne
Arene (aromatic ring)		None	 Benzene
Halide	 (X = F, Cl, Br, I)	None	CH ₃ Cl Chloromethane
Alcohol		-ol	CH ₃ OH Methanol
Ether		ether	CH ₃ OCH ₃ Dimethyl ether
Monophosphate		phosphate	CH ₃ OPO ₃ ²⁻ Methyl phosphate
Diphosphate		diphosphate	CH ₃ OP ₂ O ₆ ³⁻ Methyl diphosphate
Amine		-amine	CH ₃ NH ₂ Methylamine
Imine (Schiff base)		None	 CH ₃ C(=NH)CH ₃ Acetone imine
Nitrile		-nitrile	CH ₃ C≡N Ethanenitrile
Thiol		-thiol	CH ₃ SH Methanethiol

*The bonds whose connections aren't specified are assumed to be attached to carbon or hydrogen atoms in the rest of the molecule.

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Name	Structure*	Name ending	Example
Sulfide		<i>sulfide</i>	CH ₃ SCH ₃ Dimethyl sulfide
Disulfide		<i>disulfide</i>	CH ₃ SSCH ₃ Dimethyl disulfide
Sulfoxide		<i>sulfoxide</i>	CH ₃ S ⁺ CH ₃ Dimethyl sulfoxide
Aldehyde		<i>-al</i>	CH ₃ CHO Ethanal
Ketone		<i>-one</i>	CH ₃ COCH ₃ Propanone
Carboxylic acid		<i>-oic acid</i>	CH ₃ COOH Ethanoic acid
Ester		<i>-oate</i>	CH ₃ COOCH ₃ Methyl ethanoate
Thioester		<i>-thioate</i>	CH ₃ CSCH ₃ Methyl ethanethioate
Amide		<i>-amide</i>	CH ₃ CONH ₂ Ethanamide
Acid chloride		<i>-oyl chloride</i>	CH ₃ COCl Ethanoyl chloride
Carboxylic acid anhydride		<i>-oic anhydride</i>	CH ₃ COOCH ₃ Ethanoic anhydride

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