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- Anabolism
 - Building molecules for storing energy
- Catabolism
 - Breaking down molecules for usable energy
- Metabolism
 - The combined processes of catabolism and anabolism
- Entropy
 - A measure of disorder in a system
- Kinetic Energy
 - Energy of movement
- Potential Energy
 - Stored Energy
- ATP
 - Adenosine Triphosphate, energy currency used by the body
- Energy coupling
 - A process by which an exergonic reaction provides energy to be used in an endergonic reaction, i.e. electron transport and oxidative phosphorylation

• NADH

- Electron carrier used in cellular metabolism
- NADPH
 - Electron carrier used by chloroplasts in photosynthesis
- Hydrolysis
 - Breaking down a polymer by inserting a water molecule
- Active Site
 - The portion of an enzyme to which a substrate binds and where reactions take place.
- Substrate
 - What is acted on by an enzyme
- Product
 - What is produced by an enzyme from a substrate
- Coenzyme
 - A non-protein organic substance that combines with a protein to form an active enzyme.
- Cofactor
 - Non-protein molecules required for enzymes to function.
 - Typically refers to a non-organic molecule

- Energy of Activation
 - The energy required to initiate a reaction
- Competitive Inhibitor
 - An inhibitor that competes for the active site on an enzyme
- Noncompetitive Inhibitor
 - An inhibitor that does not compete for the active site of an enzyme
- Allosteric Inhibitor
 - An inhibitor that binds to the allosteric site of an enzyme. Results in change of shape to the active site.
- Feedback Inhibition
 - The product of an enzymatic reaction is an allosteric inhibitor of the enzyme that creates it
- Exergonic Reactions
 - Release heat
- Endergonic Reactions
 - Absorb heat

- Diffusion
 - Movement of a solute from high concentration to low concentration
- Osmosis
 - Diffusion of water
- Hypertonic
 - More solute outside than inside
- Hypotonic
 - Less solute outside than inside
- Isotonic
 - Same amount of solute outside as inside
- Freeze Fracture
 - Technique used to prove the fluid mosaic model of cell membranes
- Induced Fit
 - When a substrate binds to an enzyme, the enzyme's shape is slightly altered.

• Plasmolysis

- The removal of water from plant cells, caused by being in a hypertonic solution
- Shrinkage or contraction of the protoplasm away from the wall of a living plant or bacterial cell.

Hemolysis

- The lysis of red blood cells, caused by placing them into a hypotonic solution
- Crenation
 - The shrinkage of red blood cells when placed in a hypertonic solution
- Turgor Pressure
 - The pressure created when the central vacuole in a plant cell fills with water, allows plants to stand upright. Form of hydrostatic pressure.
- Symport
 - The active transport of substance A drives the active transport of substance B in the same direction
- Antiport
 - The active transport of substance A drives the active transport of substance B in the opposite direction
- Passive Transport
 - The movement of substances that does not require energy
- Active Transport
 - The movement of substances requiring energy (ATP)

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- Receptor Mediated Endocytosis
 - Endocytosis triggered by the activation of a receptor on the cell surface
- Exocytosis
 - Ejecting something from the cell
- Phagocytosis
 - "Cell Eating" used to bring in large molecules, usually other cells
- Pinocytosis
 - "Cell Drinking" used to bring in smaller, non-visible, water-soluble molecules
- Cyclic Phosphorylation
 - More ATP than NADPH are needed to fix carbon in Calvin cycle
- Non-cyclic phosphorylation
 - Electron Transport Chain used in photosynthesis to make ATP
- Anaerobic
 - Does not use oxygen
- Aerobic
 - Uses Oxygen

Stroma

- The space inside the chloroplast
- Thylakoid
 - Small, stacked, round discs inside the chloroplast
- Photosystem II
 - First light gathering system in non-cyclic phosphorylation
- Photosystem I
 - Second light gathering system in non-cyclic phosphorylation, after electron transport chain
- First Law of Thermodynamics
 - Energy in the universe is constant or total quantity of energy before equals total quantity after each energy conversion.
 - ENERGY can be transferred and transformed but energy cannot be created or destroyed.
- Second Law of Thermodynamics
 - Every **energy** transfer makes the universe more disordered, or increases entropy

	High to low or low to high concentration	Requires a protein (yes or no)	Requires added energy (yes or no)
Diffusion			
Facilitated transport = facilitated diffusion			
Active transport			

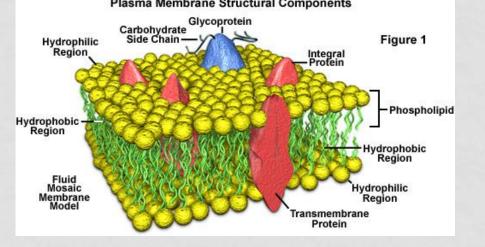
	High to low or low to high concentration	Requires a protein (yes or no)	Requires added energy (yes or no)
Diffusion	High to Low	No	No
Facilitated transport = facilitated diffusion	High to Low	Yes	No
Active transport	Low to High	Yes	Yes

	Red Blood Cell	Plant cell
Hypertonic		
Isotonic		
Hypotonic		

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	Red Blood Cell	Plant cell
Hypertonic	Crenation	Plasmolysis
Isotonic	Nothing	Nothing
Hypotonic	Hydrolysis	Turgor Pressure

- What is the fluid mosaic model for cell membranes?
 - Phospholipids and membrane proteins flow (fluid) and membrane proteins are embedded in the membrane (mosaic)
- What is the structure of a membrane (describe and draw it and include hydrophilic and hydrophobic labels, plus phospholipids, integral proteins, peripheral proteins and glycoproteins)?
 Plasma Membrane Structural Components



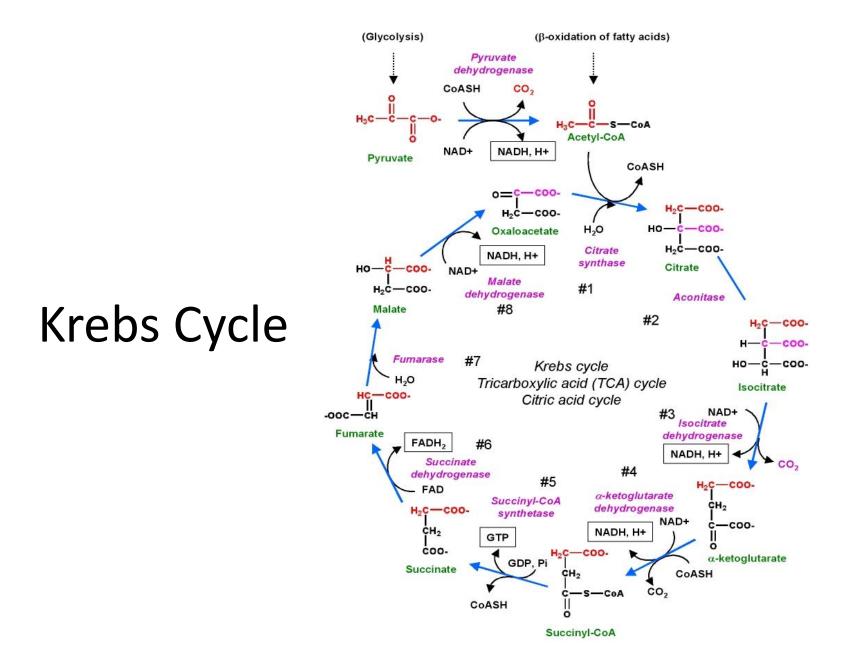
- What is the fluid mosaic model for cell membranes?
 - Phospholipids and membrane proteins flow (fluid) and membrane proteins are embedded in the membrane (mosaic)
- What is the structure of a membrane (describe and draw it and include hydrophilic and hydrophobic labels, plus phospholipids, integral proteins, peripheral proteins and glycoproteins)?
 - See previous slide
- What are some of the functions of plasma membrane integral proteins?
 - Transport, Recognition, Intercellular Joining, Attachment
- What molecules can pass through a cell membrane without the aid of a protein? What molecules require the aid of a protein?
 - Without Protein: Small nonpolar molecules lipids
 - With Protein: Small charged particles, polar substances, larger molecules ions, proteins

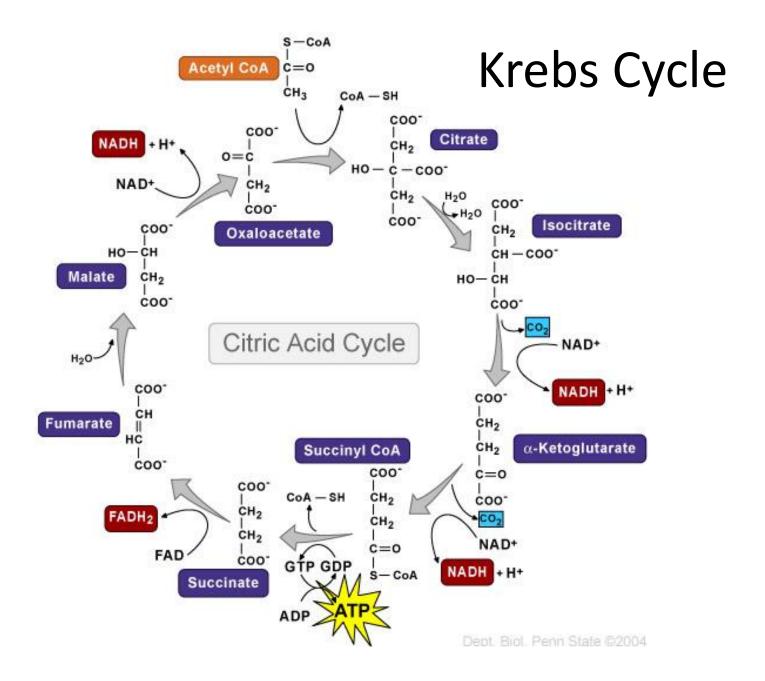
- How do enzymes lower the energy of activation (or speed up a chemical reaction)?
 - They allow substrates to react with each other faster by straining the reactant's bonds and bringing the reactants closer together.
- Why do enzymes have an optimal pH? Why does pepsin from the stomach act at a lower pH than proteases from the small intestine?
 - pH effects the structure of a protein. Proteins will denature outside the range in which they are designed to work.
- Why do enzymes have an optimal temperature? Why do bacteria from hot springs have a higher optimal temperature than human enzymes?
 - Proteins denature at temperatures outside their optimal range.

	CO2 per glucose	Oxygen consumed (yes/no)	Oxygen required (yes/no)	Initial step	Final step	Where proteins located
Glycolysis						
Transition Step						
Krebs cycle						
Electron Transport						
Oxidative phosphoryl ation (ATP synthase)						

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FILL IN THE BLANK						
	CO2 per glucose	Oxygen consumed (yes/no)	Oxygen required (yes/no)	Initial step	Final step	Where proteins located
Glycolysis	0	no	no	Glucose 2 ATP	2 Pyruvate 4 ATP 2 NADH	Cytoplasm
Transition Step	2	no	yes	2 Pyruvate	2 Acetyl CoA 2 NADH	Cytoplasm
Krebs cycle	4	no	no	2 Acetyl CoA 2 Oxaloacetate	6 NADH 2 FADH2 2 ATP	Mitochondrial Matrix
Electron Transport	0	yes	yes	10 NADH 2 FADH	34 ATP	Inner Mitochondrial Membrane
Oxidative phosphoryl ation (ATP synthase)	6	yes	yes	Glucose	36-38 ATP	Cytoplasm/ Mitochondria





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- How many ATP are made through cellular respiration (from one glucose to CO₂ and H₂O)?
 - 36-38
- How many ATP are made from one NADH?
 3
- How many ATP are made from one FADH?
 - 2
- Where is ATP synthase located?
 - In the inner mitochondrial membrane
- Compare where mitochondria and chloroplasts put the H+ ions.
 - They both put them into a small space to better form a concentration gradient (intermembrane space in mitochondria and thylakoid space in chloroplasts)

- Why are plants green?
 - The chlorophyll in plants absorbs all spectra of light except green
- Where does oxygen come from in photosynthesis?
 - Water
- Which pigments are responsible for absorbing light energy for the light dependent reactions?
 - Chlorophylls
- Where are the photosynthetic pigments located within the chloroplast?
 - In the thylakoid membrane

•Know the key steps (organic molecules) of the Calvin-Benson cycle, especially initial steps and final steps. Which molecule forms glucose?

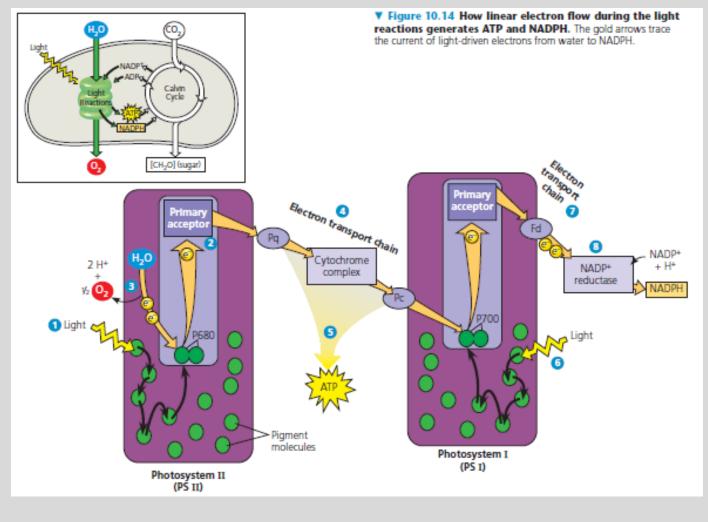
•Carbon Dioxide

•What are the differences between "light dependent reactions" of photosynthesis and Calvin Cycle or "light independent reactions" of photosynthesis? [Quick summary table]

Process	Location	Reactants	Products
Light dependent reactions			
Light Independent reactions or Calvin Cycle			

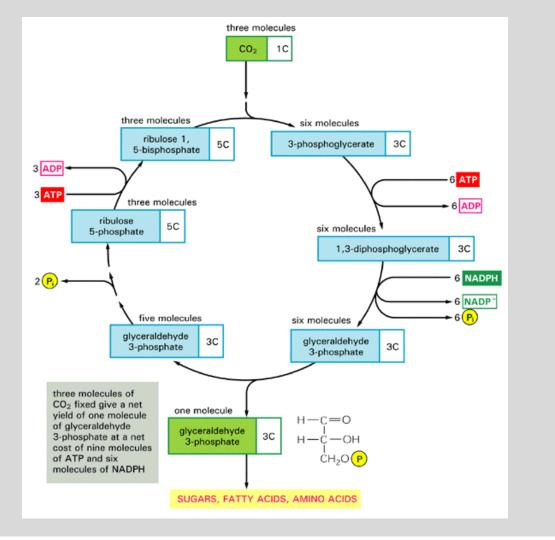
Process	Location	Reactants	Products	
Light dependent reactions	Thylakoid Membrane	H ₂ 0 NADP+ ADP Light	Oxygen ATP NADPH	
Light Independent reactions or Calvin Cycle	Stroma	3 CO ₂ 9 ATP 6 NADPH	2 G3P 9 ADP, 6 NADP+.	

Light Dependent Reactions



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Calvin Cycle



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Questions



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